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In chemistry, the term "bent" can be applied to certain molecules to describe their molecular geometry. Certain atoms, such as oxygen, will almost always set their two (or more) covalent bonds in non-collinear directions due to their electron configuration. Water (H_2O) is an example of a bent molecule, as well as its analogues. The bond angle between the two hydrogen atoms is approximately 104° ...

Bent molecular geometry - Wikipedia

In chemistry, trigonal planar is a molecular geometry model with one atom at the center and three atoms at the corners of an equilateral triangle, called peripheral atoms, all in one plane. In an ideal trigonal planar species, all three ligands are identical and all bond angles are 120° . Such species belong to the point group D_{3h} . Molecules where the three ligands are not identical, such as H_2O ...

Trigonal planar molecular geometry - Wikipedia

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Chapter 9, Section 1 Monarchs, Nobles, and the Church During feudal times, monarchs in Europe stood at the head of society but had limited power.

Chapter 9

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